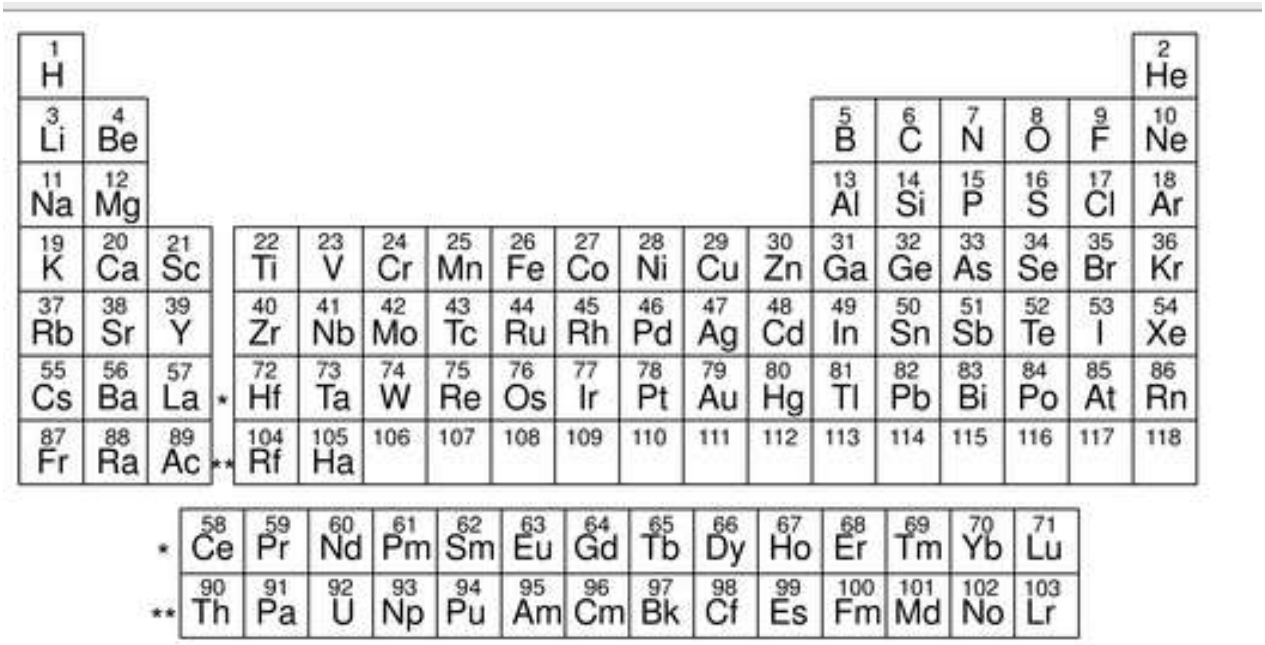
UNIT 2 lecture 2 Periodic Table HW Name:_____Period:_____

On the periodic table label the following:

- groups (1A to 8A)
- periods (1 to 7)
- groups with special names (lanthanides, halogens, alkaline earth metals, alkali metals, noble gases, actinide, transition metals)
- staircase (draw it in and then fill in the metals blue, the non-metals yellow and the metalloids green



Fill in the following table:

element	Element symbol	atomic mass	atomic number	metal,non-metal, or metalloid	period	group (special name if one of the four we discussed)
Xenon						
Astatine						
Copper						
Radium						
Calcium						
Lithium						

Which of the Above elements would have similar properties to each other? WHY?